## Material Safety Data Sheet

## CALCIUM CHLORIDE

|  | I.- | Product Identification |
| :--- | :--- | :--- |
| Manufacturer | $:$ | Industria Del Alcali, S. A. De C. V. |
| Address | $:$ | Carretera a García Km. 9 |
|  |  | García, N L. |
| Telephone | ( 81 )8863-26-00 |  |
| Chemical Name | $:$ | Calcium Chloride |
| Chemical Family | $:$ | Alkaline Salts |
| Chemical Formula | $: \mathrm{CaCl}_{2}$ |  |

## II .- Dangerous Ingredients

It does not present noxious substances


Effective: February - 2005
Replace : February-2003

## CALCIUM CHLORIDE

## V.- Fire And Explosion Information

Calcium Chloride is a NON - FLAMMABLE chemical product.
It is not susceptible to produce flames or explosions.

## VI .- Reactivity Data

Stability $\quad:$ It is a stable product at normal conditions.
Incompatibility $\quad:$ It is highly hygroscopic, releasing heat when solved in water.

## VII.- Cautions

It is not absorbed through skin.
In case of contact : Wash immediately with running water, for at least fifteen minutes
with the eyes Call a doctor or take patient to the nearest medical facility.

In case of contact : Wash with water until it is clean.
with the skin
In case of ingestion : Immediately induce vomiting giving the patient two glasses of water and introduce finger in throat.

Inhalation
: Take patient to an area with fresh air.

## CALCIUM CHLORIDE

## VIII.- Personal Safety Equipment

During the handling and use of Calcium Chloride it is convenient to protect :
a).- The respiratory system, by using a breathing mask with dust filter
b).- The eyes, using safety glasses or goggles.
c).- The skin, using gloves, and apron which does not permit alkaline solutions to pass through.

## IX.- Spills And Wastes

If the Calcium Chloride is spilled, it should be picked up with a shovel, filling a receptacle properly labeled and closed for later disposal.

Wash with plenty of water and contain to avoid its flow through water ways and the sewage system. Ask for assistance for its disposal.

The Calcium Chloride 94 \% solution is neutralized by adding Sodium Carbonate, obtaining a mixture of Calcium Carbonate and Sodium Chloride.

Calcium Chloride waste should be handled according to the actual State and / or Federal regulation.

